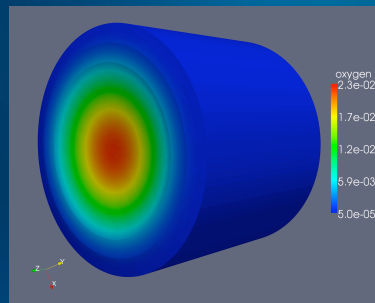


A Strategy for Simulation of Long Term Fuel Storage and Subsequent Transport

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Fuel lifetime

Fuel Performance

- Hydrogen uptake
- Cladding abuse (fretting)

Drying

- Hydride reorientation
- Thermal soak
- High stresses

Long term storage

- Corrosion (?)
- Creep (?)
- Fuel oxidation (?)

Transport

- Low cycle fatigue
- Fracture toughness

Off-normal events

- Accidents
- Drop simulation

- Fuel performance calculation is the basis for cladding stress state on entry to drying cycle
- Not sure what the relevant physics is across the cycle
 - Rapid “plug and play” model replacement to support exploration of various effects
 - Sensitivity analysis
- How does longer burnup effect aging?
- How do all these uncertainties add up?

Simulation can answer many questions

- Easily look at the complete cycle, different fuel and reactor types, different operational conditions (longer burnup)
- Readily build intuition and test relative importance of different phenomena
- Employ implicit integration to allow 100 year storage periods to be computed with only a few time steps.
- Get results quickly and relatively inexpensively.
- Track predictions and address unexpected issues in long term tests

Assist with analytically closing the gaps.

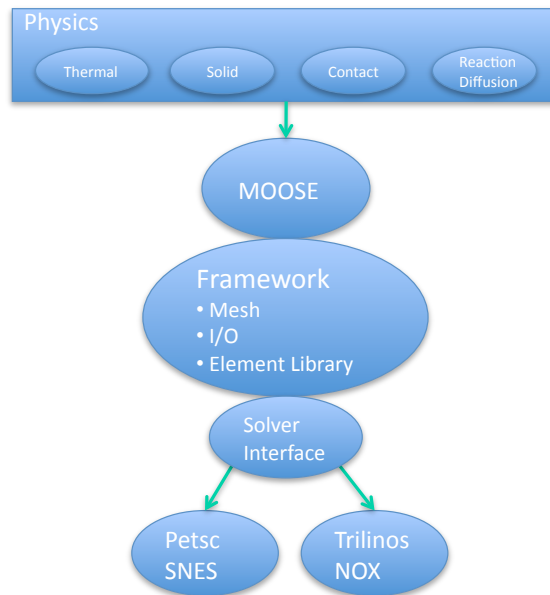
Simulation strategy

- Develop simulation basis for long term storage and subsequent transport
- Fuel performance code sets initial conditions for drying
- Drying process modeled with same code, using hydride and thermomechanical models in cladding (fuel still present for PCI).
- Storage period may include corrosion (inside the fuel) and cladding creep.
- Transport will include vibration, stress and strain, and low cycle fatigue in cladding

What capabilities can be leveraged to perform end-to-end calculations of this sort?

MOOSE: A coupled multiphysics framework

- Clean physics interface for ease of development of new applications
 - Physics Interface conceals framework complexity
- Framework provides core set of common services
 - libMesh: <http://libmesh.sf.net>
- 1D, 2D, and 3D; steady state and transient with same code
- Adaptive, parallel, fully coupled, fully implicit, portable
 - Robust solvers are key for “ease of use”
 - Load balancing and mesh adaptation are tightly integrated to form a robust “dial free” application



MOOSE Ecosystem

Application	Physics	Start	Time To Results
BISON	Thermo-mechanics, Chemical Diffusion, coupled mesoscale	June 2008	4 Months
PRONGHORN	Neutronics, Porous Flow, Eigenvalue	September 2008	3 Months
SALMON	Multiphase Porous Flow	June 2009	3 Months
MARMOT	4 th Order Phasefield Mesoscale	August 2009	1 Month
RAT	Porous ReActive Transport	August 2009	1 Month
FALCON	Geo-mechanics, coupled mesoscale	September 2009	3 Months

BISON – A MOOSE fuel performance code

- Transient, nonlinear heat conduction

$$\rho C_p T_t - \nabla \cdot k \nabla T - q = 0$$

- Nonlinear oxygen diffusion

$$s_t - \nabla \cdot \left(D \left(\nabla s + \frac{s Q^*}{F R T^2} \nabla T \right) \right) = 0$$

- Linear elastic model, nonlinear material properties

$$\begin{aligned} (u_{tt}, \phi) &+ \mu S(u, \phi) + \lambda (\nabla \cdot u, \nabla \cdot \phi) \\ &- (f, \phi) - \langle g, \phi \rangle - (\alpha T, \nabla \phi) = 0 \end{aligned}$$

$$S(u, \phi) = \sum_{i,j=1}^3 (\partial_j u_i + \partial_i u_j) (\partial_j \phi_i + \partial_i \phi_j)$$

Additional physics currently in BISON

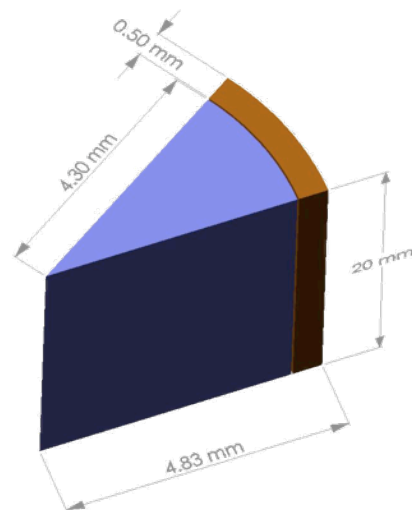
- Temperature and burnup dependent thermal properties of fuel
- Fission product swelling
 - Gas
 - Solid
- Burnup
- Fission gas release in progress
- Other closure relations

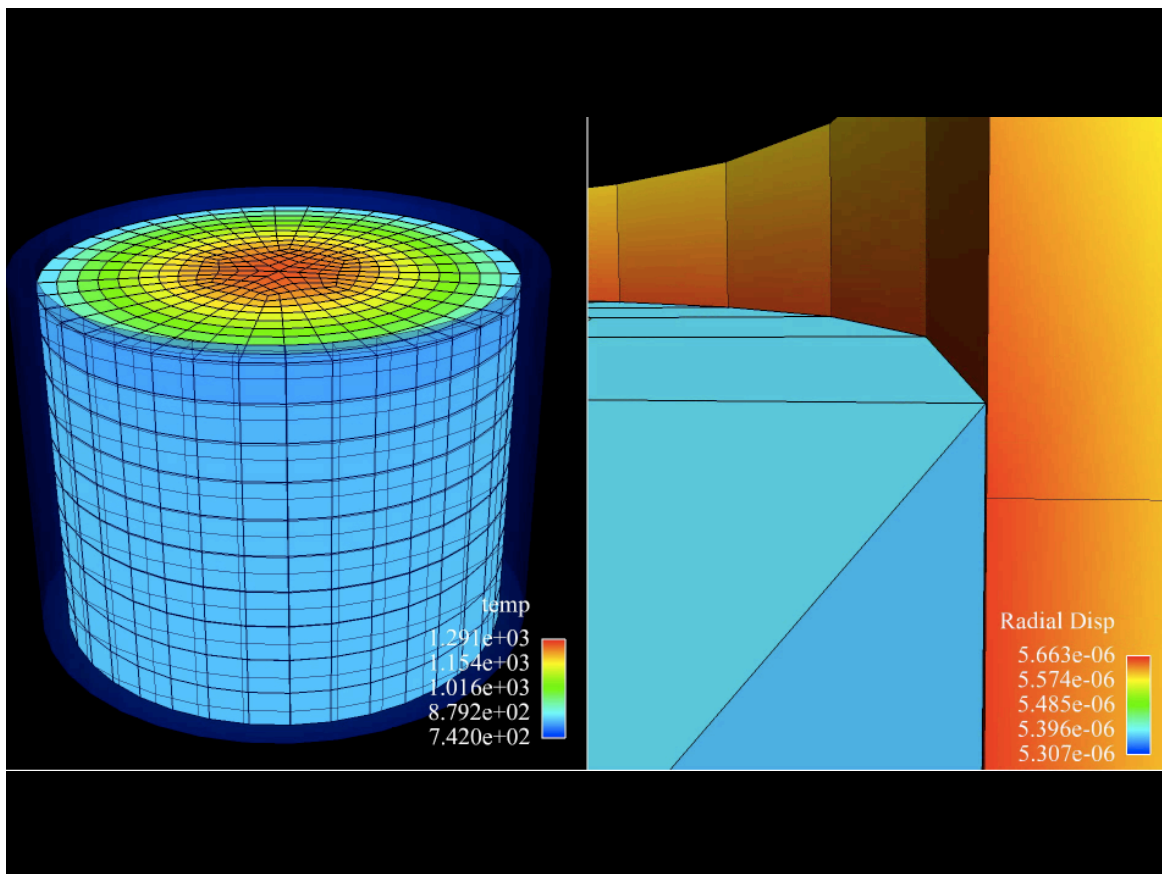
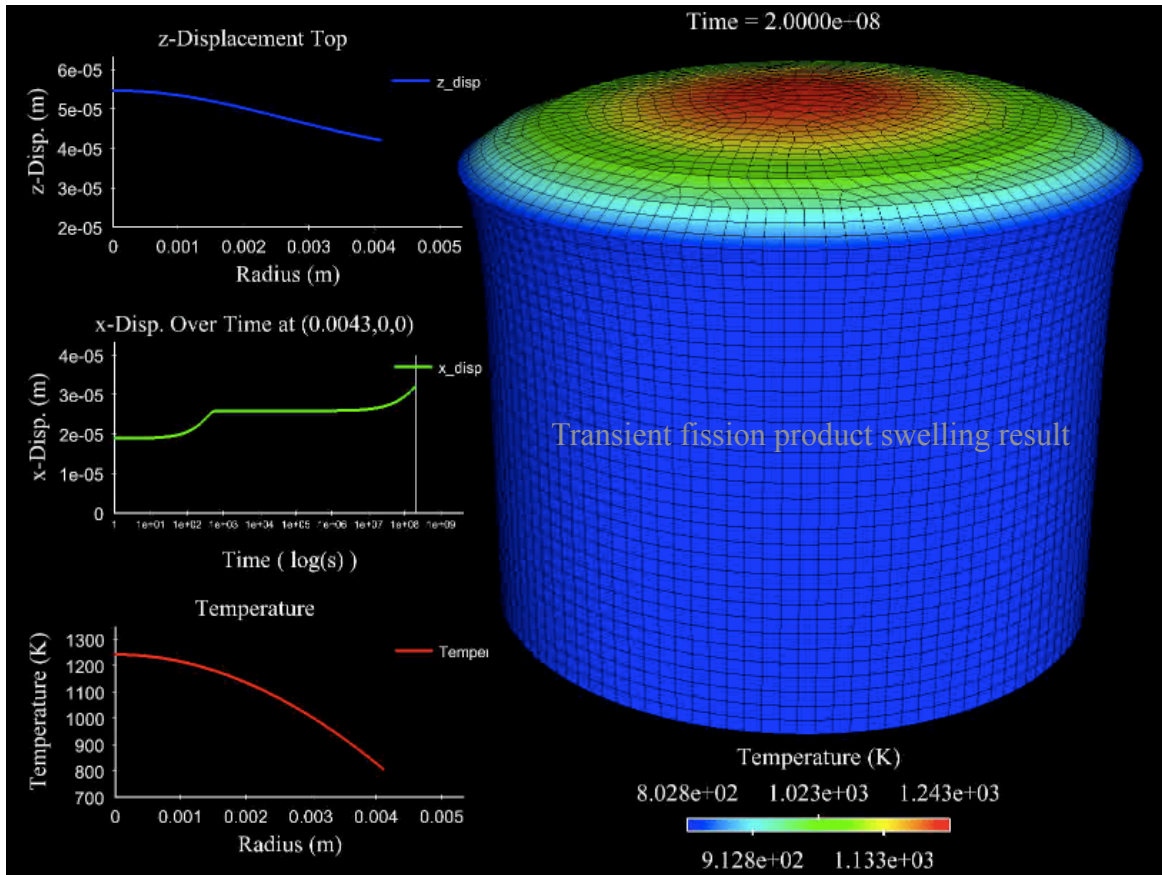
Potential Collaboration

- ANATECH has expressed interest in active collaboration on MOOSE/ BISON development
- Joe and Mark Rashid have done significant recent work in damage mechanics and hydride reorientation
- FALCON includes a spent fuel capability including post-irradiation cladding creep

Coupled thermomechanics & oxygen diffusion

- Extension of previous work of Ramirez, Stan, and Cristea, J. Nuclear Materials, 2006.
- Extended to 3D
- JFNK solution of fully coupled thermomechanics and oxygen diffusion

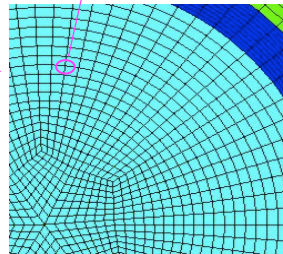
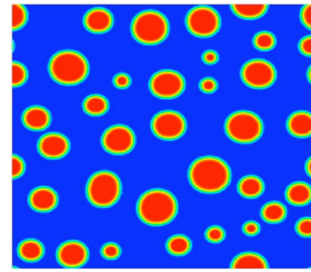
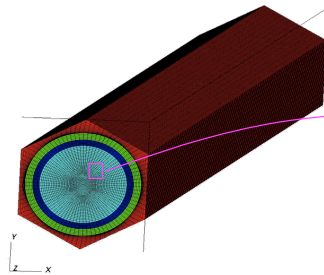




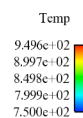
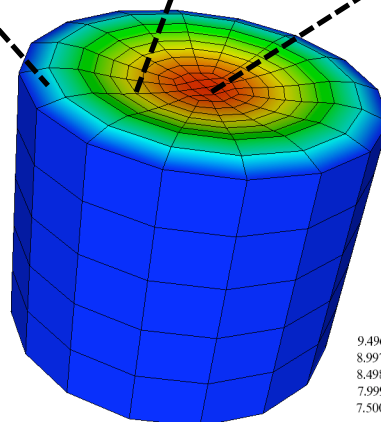
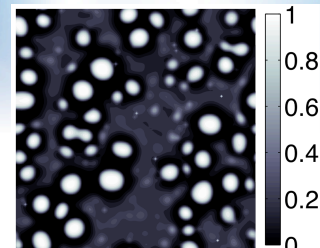
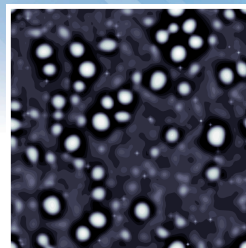
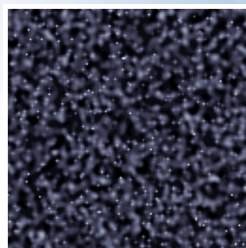
Fully-coupled bridging to/from mesoscale

The energy equation is solved at the engineering scale for temperature using JFNK

$$\rho C_p \frac{\partial T}{\partial t} = \nabla \cdot q + \dot{Q}$$



S.K. Rokkam, P.C. Millett, D. Wolf, and A. El-Azab, "Phase Field Simulation of Void Growth in Irradiated Materials," Fourth International on Multiscale Materials Modeling, Symposium 4-Multiscale modeling of microstructure evolution in materials, Oct. 27-31, pages 405-408



Equation System

- All variables are solved simultaneously
 - Phase field residual equations

$$\mathbf{R}_{c_i} = \frac{\partial c_i}{\partial t} - \nabla \cdot \left(M_{ij} \nabla \left(\frac{\partial g_0}{\partial c_i} - \kappa \nabla^2 c_i + \frac{\partial E_{el}}{\partial c_i} \right) \right) = 0 \quad \mathbf{R}_{\eta_i} = \frac{\partial \eta_i(\mathbf{r}, t)}{\partial t} + L_i \left(\frac{\partial f_0}{\partial \eta_i} - \kappa \nabla^2 \eta_i + \frac{\partial E_{el}}{\partial \eta_i} \right) = 0$$

- Coupled variable residual equations

$$\mathbf{R}_u = \nabla \cdot (\mathbf{C} \nabla \mathbf{u}) - \nabla \cdot (\mathbf{C} \boldsymbol{\varepsilon}^*) = 0$$

- FEM discretization

$$c_i(\mathbf{r}) = \sum_{j=1}^N c_i^j \varphi_j(\mathbf{r})$$

Discretized using 3rd order Hermite element

2D: 20 DOF

3D: 36 DOF

$$\eta_i(\mathbf{r}) = \sum_{j=1}^N \eta_i^j \varphi_j(\mathbf{r})$$

Discretized using 1st order Lagrange elements

2D: 8 DOF

3D: 12 DOF

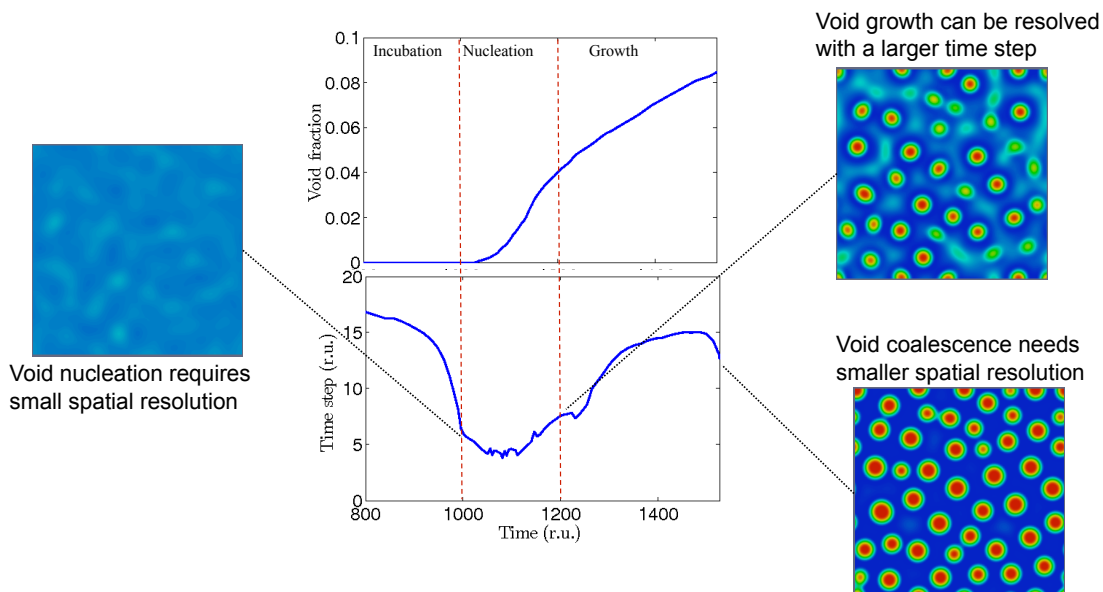
$$\mathbf{u}(\mathbf{r}) = \sum_{j=1}^N \mathbf{u}^j \varphi_j(\mathbf{r})$$

Discretized using 1st order Lagrange elements

2D: 8 DOF

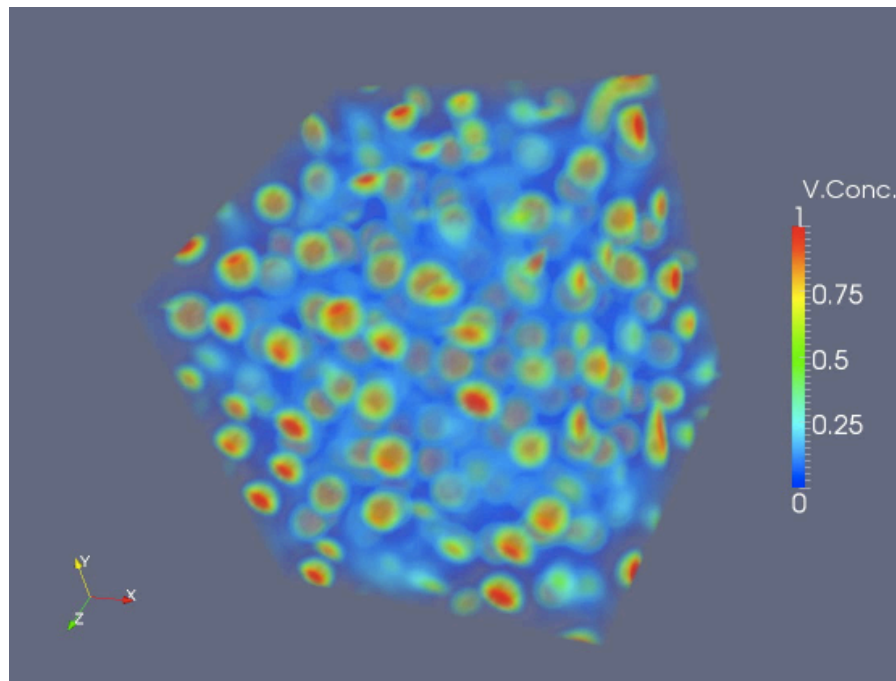
3D: 12 DOF

Adaptive Time Step



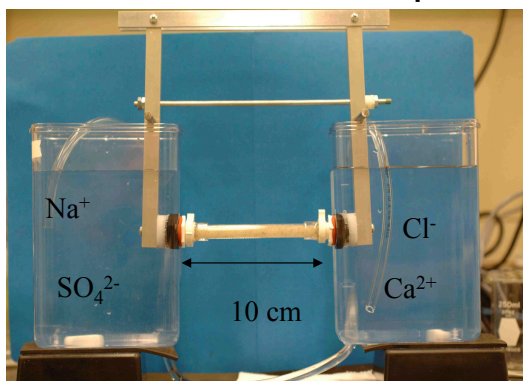
- Solution time step adapts to the driving phenomena

3-D Void Nucleation and Growth

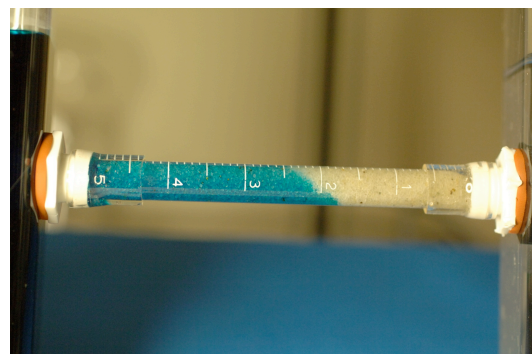


Modeling of coupled diffusion and mineral precipitation in porous media

Double Diffusion-Reaction experiment



Focused precipitation front – visible only by using dye



$$\begin{aligned}
 (1) \quad & \frac{\partial \left[\theta \left(C_{Ca^{2+}} + C_{CaCl^+} + C_{CaCl_2(aq)} + C_{CaOH^+} + C_{CaSO_4(aq)} + C_{CaSO_4(s)} \right) \right]}{\partial t} - \nabla \cdot \left[\theta D \cdot \nabla \left(C_{Ca^{2+}} + C_{CaCl^+} + C_{CaCl_2(aq)} + C_{CaOH^+} + C_{CaSO_4(aq)} \right) \right] = 0 \\
 (2) \quad & \frac{\partial \left[\theta \left(C_{Cl^-} + C_{CaCl^+} + 2C_{CaCl_2(aq)} + C_{HCl(aq)} + C_{NaCl(aq)} \right) \right]}{\partial t} - \nabla \cdot \left[\theta D \cdot \nabla \left(C_{Cl^-} + C_{CaCl^+} + 2C_{CaCl_2(aq)} + C_{HCl(aq)} + C_{NaCl(aq)} \right) \right] = 0 \\
 (3) \quad & \frac{\partial \left[\theta \left(C_{H^+} + 2C_{H_2SO_4(aq)} + C_{HCl(aq)} + C_{HSO_4^-} - C_{CaOH^+} - C_{NaOH(aq)} - C_{OH^-} \right) \right]}{\partial t} - \nabla \cdot \left[\theta D \cdot \nabla \left(C_{H^+} + 2C_{H_2SO_4(aq)} + C_{HCl(aq)} + C_{HSO_4^-} - C_{CaOH^+} - C_{NaOH(aq)} - C_{OH^-} \right) \right] = 0 \\
 (4) \quad & \frac{\partial \left[\theta \left(C_{Na^+} + C_{NaCl(aq)} + C_{NaOH(aq)} + C_{NaSO_4^-} \right) \right]}{\partial t} - \nabla \cdot \left[\theta D \cdot \nabla \left(C_{Na^+} + C_{NaCl(aq)} + C_{NaOH(aq)} + C_{NaSO_4^-} \right) \right] = 0 \\
 (5) \quad & \frac{\partial \left[\theta \left(C_{SO_4^{2-}} + C_{CaSO_4(aq)} + C_{H_2SO_4(aq)} + C_{HSO_4^-} + C_{NaSO_4^-} + C_{CaSO_4(s)} \right) \right]}{\partial t} - \nabla \cdot \left[\theta D \cdot \nabla \left(C_{SO_4^{2-}} + C_{CaSO_4(aq)} + C_{H_2SO_4(aq)} + C_{HSO_4^-} + C_{NaSO_4^-} \right) \right] = 0 \\
 (6) \quad & \frac{d(C_{CaSO_4(s)})}{dt} - 0.1 \times 6.456542 \times 10^{-8} \times \left(1 - \frac{C_{Ca^{2+}} \cdot C_{SO_4^{2-}}}{10^{-1.8487}} \right) = 0 \\
 (7) \quad & C_{CaCl^+} - 10^{-20.7} C_{Ca^{2+}} \cdot C_{Cl^-} = 0 \\
 (8) \quad & C_{CaCl_2(aq)} - 10^{-20.653} C_{Ca^{2+}} \cdot (C_{Cl^-})^2 = 0 \\
 (9) \quad & C_{CaOH^+} - 10^{-12.85} C_{Ca^{2+}} \cdot (C_{H^+})^{-1} = 0 \\
 (10) \quad & C_{CaSO_4(aq)} - 10^{-2.1} C_{Ca^{2+}} \cdot C_{SO_4^{2-}} = 0 \\
 (11) \quad & C_{H_2SO_4(aq)} - 10^{-1.021} (C_{H^+})^2 \cdot C_{SO_4^{2-}} = 0 \\
 (12) \quad & C_{HCl(aq)} - 10^{0.7} C_{H^+} \cdot C_{Cl^-} = 0 \\
 (13) \quad & C_{HSO_4^-} - 10^{1.976} C_{H^+} \cdot C_{SO_4^{2-}} = 0 \\
 (14) \quad & C_{NaCl(aq)} - 10^{-0.782} C_{Na^+} \cdot C_{Cl^-} = 0 \\
 (15) \quad & C_{NaOH(aq)} - 10^{-14.799} C_{Na^+} \cdot (C_{H^+})^{-1} = 0 \\
 (16) \quad & C_{NaSO_4^-} - 10^{0.82} C_{Na^+} \cdot C_{SO_4^{2-}} = 0 \\
 (17) \quad & C_{OH^-} - 10^{-13.991} (C_{H^+})^{-1} = 0
 \end{aligned}$$

Challenges:

Both fast and slow kinetics

Strongly coupled processes

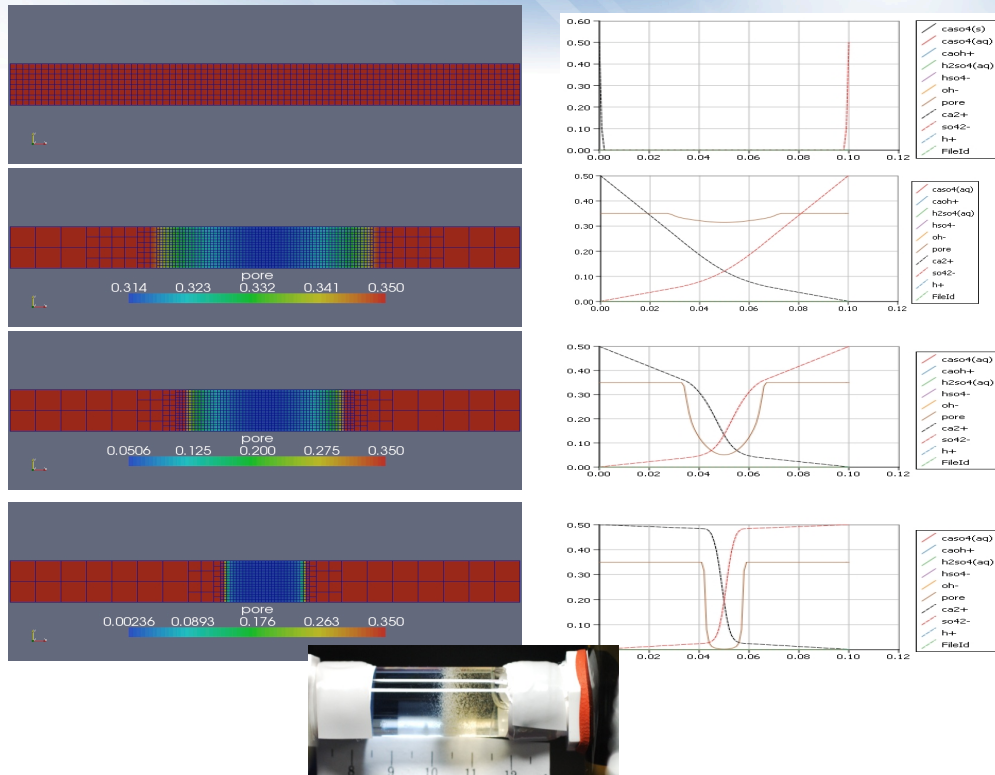
Conventional Approach:

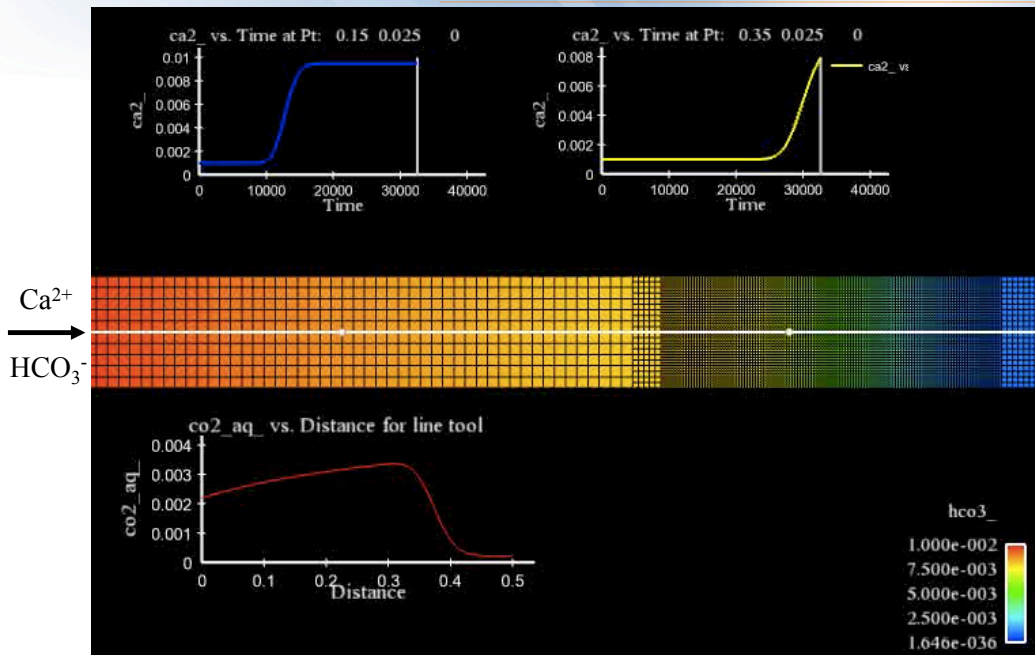
Operator splitting

Our approach:

Fully coupled, fully implicit

Adaptive mesh refinement





Concluding remarks

- The Multiphysics Object-Oriented Simulation Environment, MOOSE, supports rapid, 3-D, parallel, applications code development.
- MOOSE could readily be extended to incorporate the additional model sets that span the requirements for simulation of long term fuel storage and transport processes.